

Advanced Building Science

- Psychrometrics
 - Key aspects of the psychrometric chart
 - Using the chart to plot basic processes
 - Using the chart to solve specific problems

- Readings
 - HF: Chapter 1.1 to 1.16
 - HPE: Chapter 3.4

Pyschrometrics

- The science that deals with the physical laws governing the mixture of air and water vapor
- An applied science that involves ...
 - the properties of moist air
 - the processes in which the temperature and/or water vapor content of the mixture are changed.

Pyschrometrics

- Mixture of two non-reacting nearly ideal gases
 - dry air (molecular mass = 28.96)
 - water vapor (molecular mass = 18.01)
- It can be simply explained and is based on:
 - the ideal gas equation $\Rightarrow pV = nRT$
 - Dalton's model of partial pressures $\Rightarrow P_{TOT} = P_{DA} + P_{WV}$
 - conservation of energy
 - conservation of mass

Figure A4-6 No. 1 IP—Normal Temperature—Sea Level.



ASHRAE PSYCHROMETRIC CHART NO. 1

NORMAL TEMPERATURE

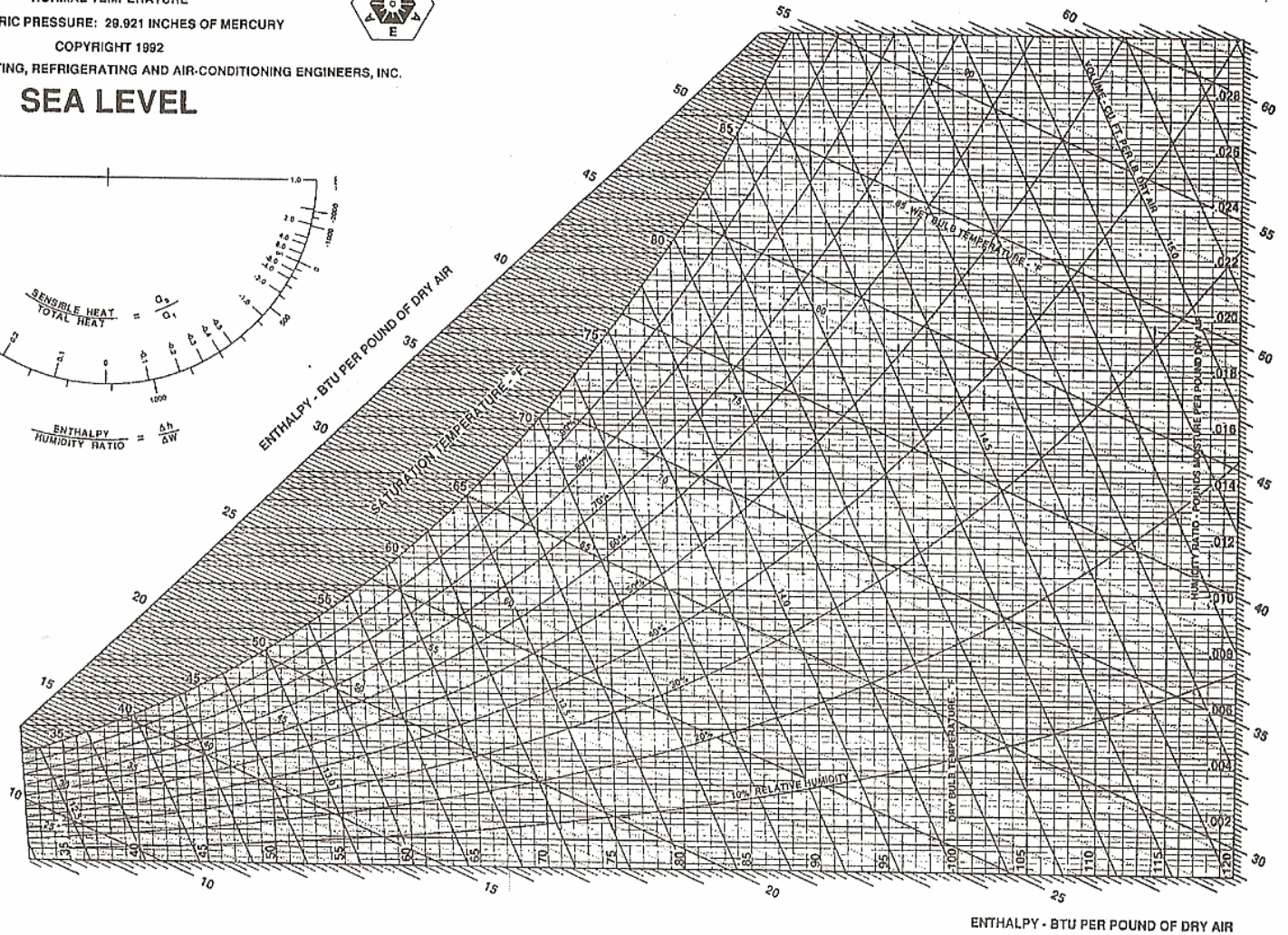
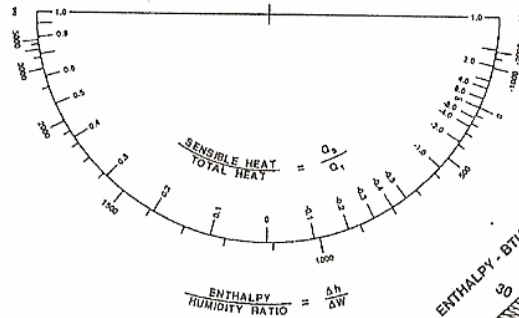
BAROMETRIC PRESSURE: 29.921 INCHES OF MERCURY

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SEA LEVEL



Source: ASHRAE Handbook Fundamentals 2009 Chapter 1

Understanding Psychrometrics

By: Donald P. Gatley

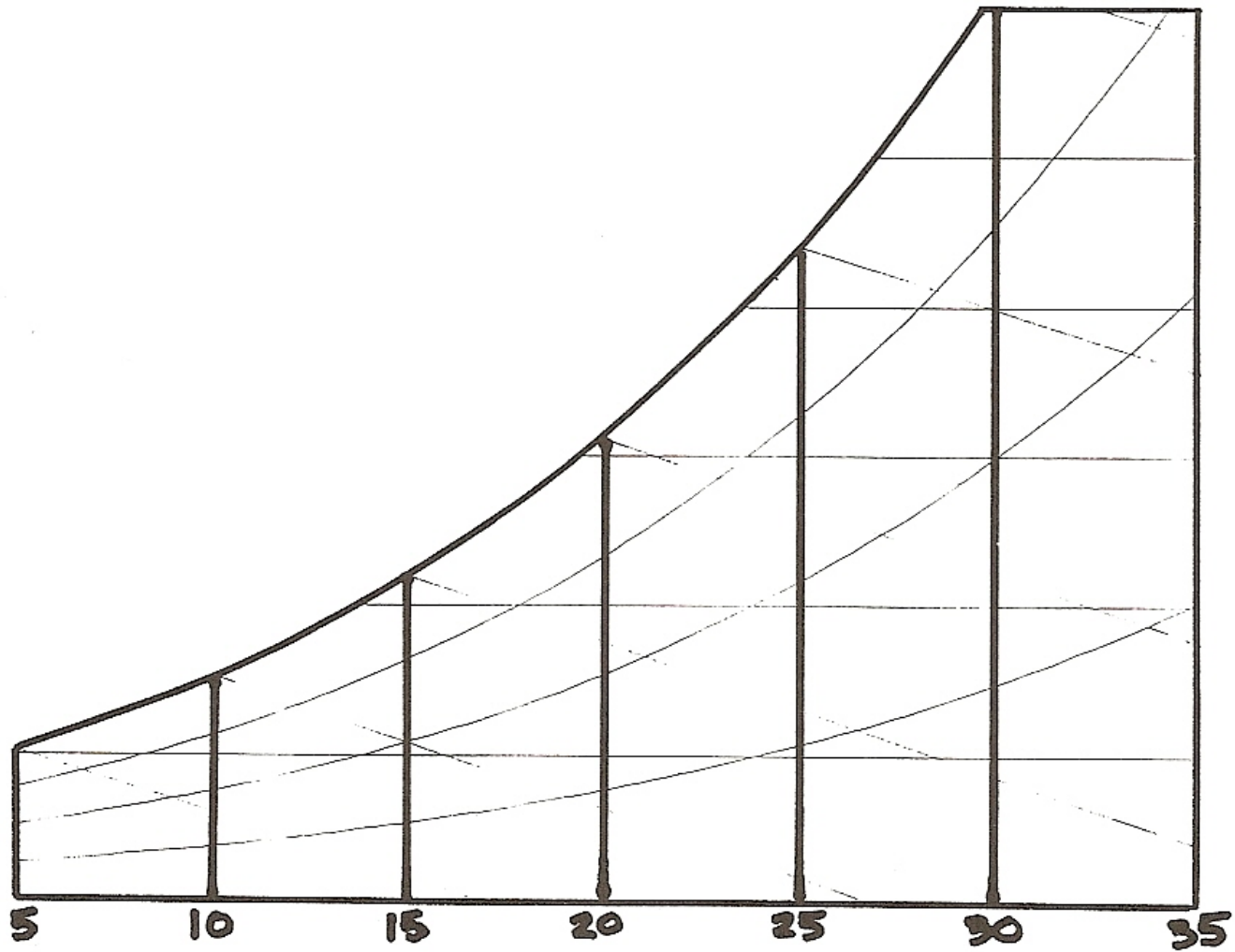


Figure 10-1 Dry-bulb temperature (t_{DB} in units of $^{\circ}C$).

Source: Gatley, Understanding Psychrometrics, Chapter 10

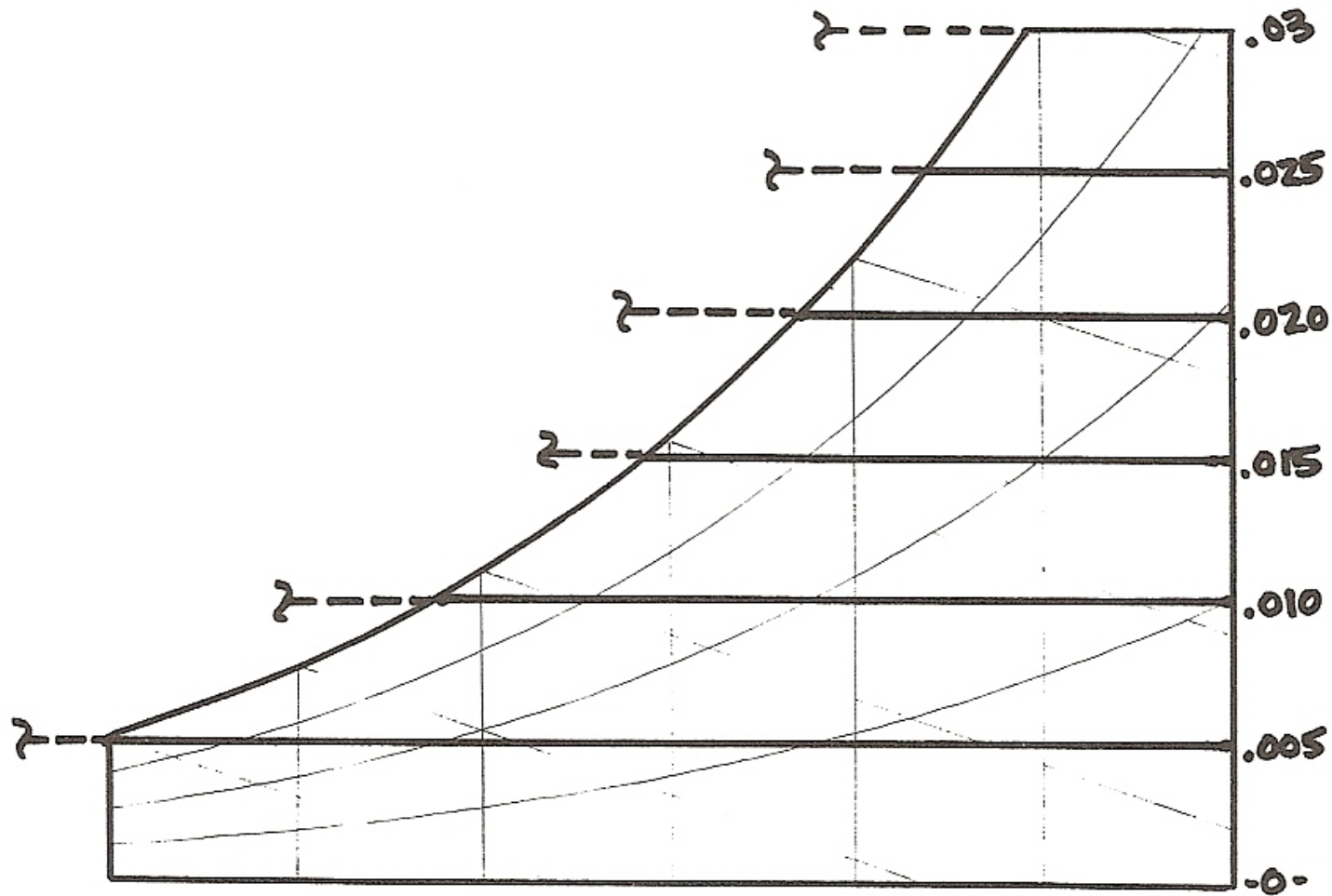


Figure 17-1 Humidity ratio isolines (W in units of $\text{kg}_{\text{WV}}/\text{kg}_{\text{DA}}$ [$\text{lb}_{\text{WV}}/\text{lb}_{\text{DA}}$]).

Source: Gatley, Understanding Psychrometrics, Chapter 17

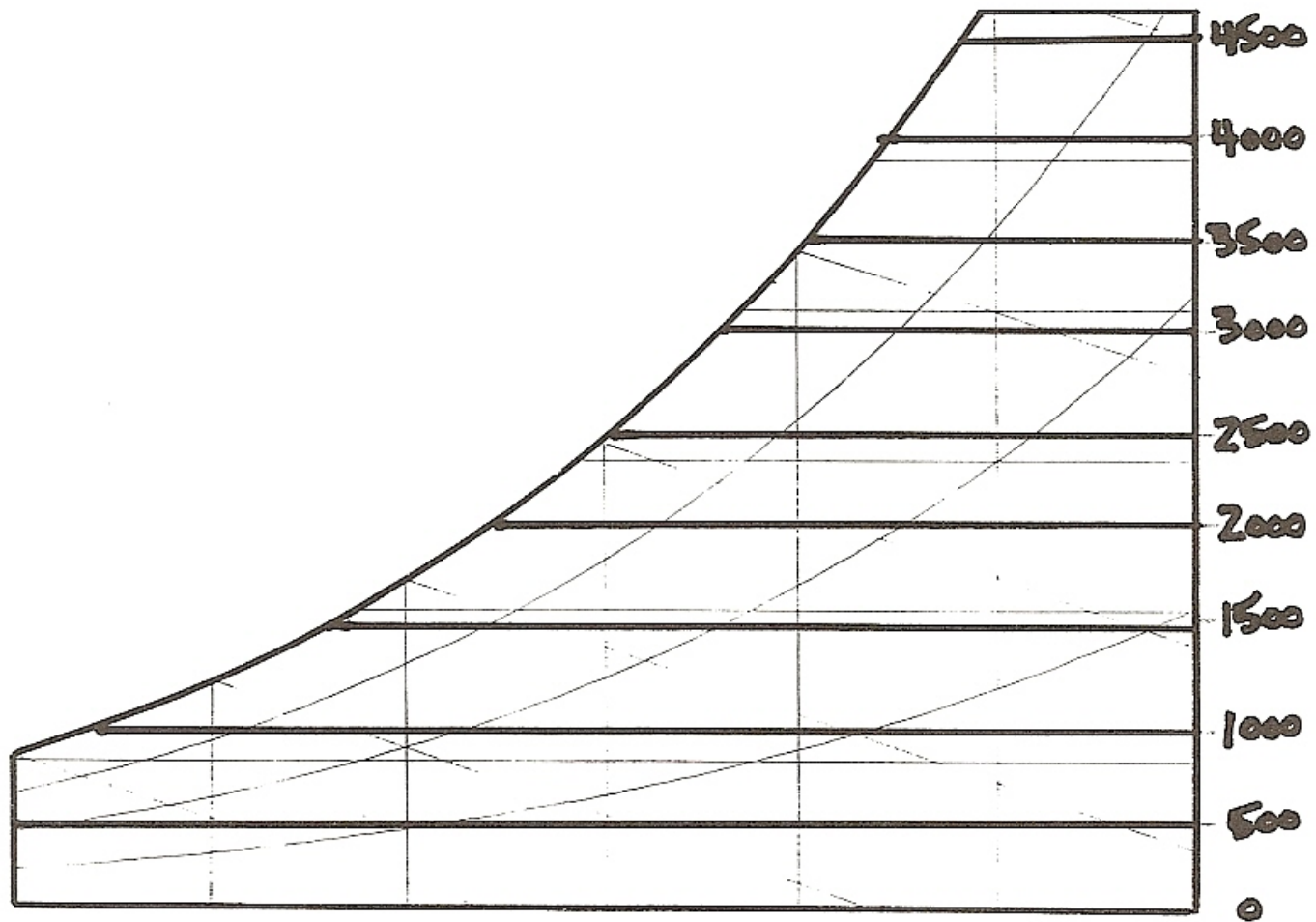


Figure 18-1 Water vapour pressure isolines (P_{WV} in units of Pa [in. Hg]).

Source: Gatley, Understanding Psychrometrics, Chapter 17

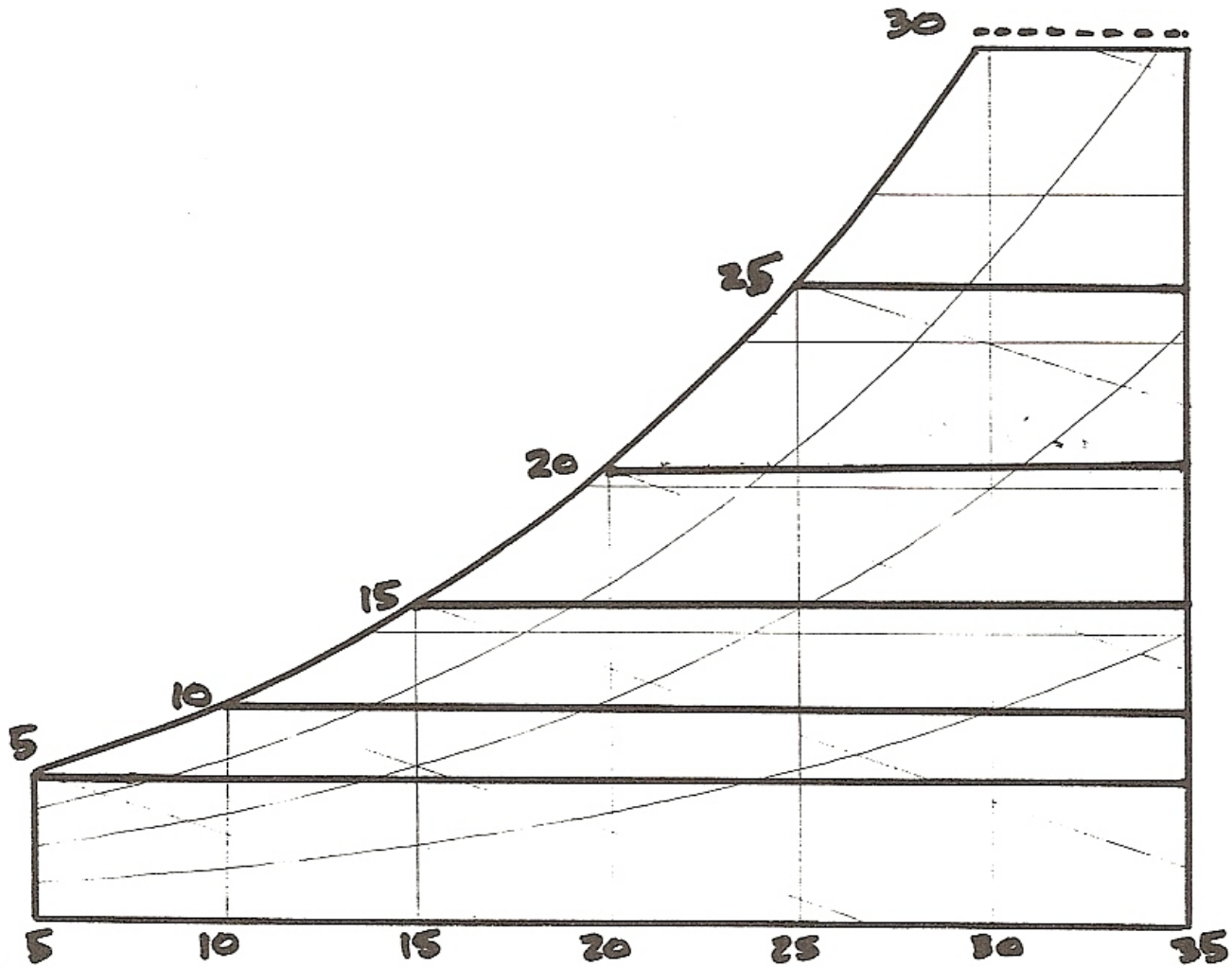


Figure 12-1 Dew-point temperature isolines (t_{DP} in units of $^{\circ}C$).

Source: Gatley, Understanding Psychrometrics, Chapter 12

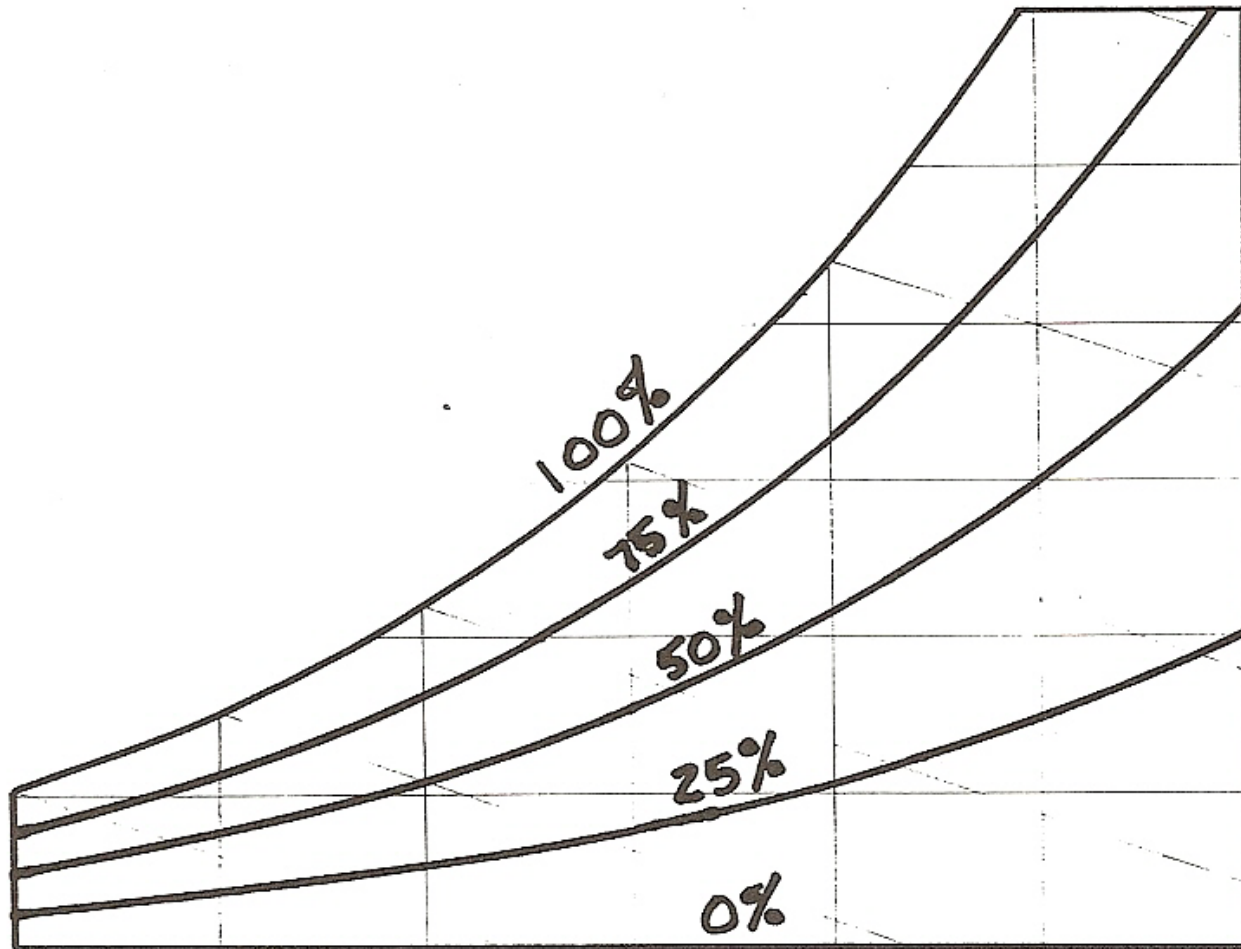


Figure 13-2 *Relative humidity isolines.*

Source: Gatley, Understanding Psychrometrics, Chapter 12

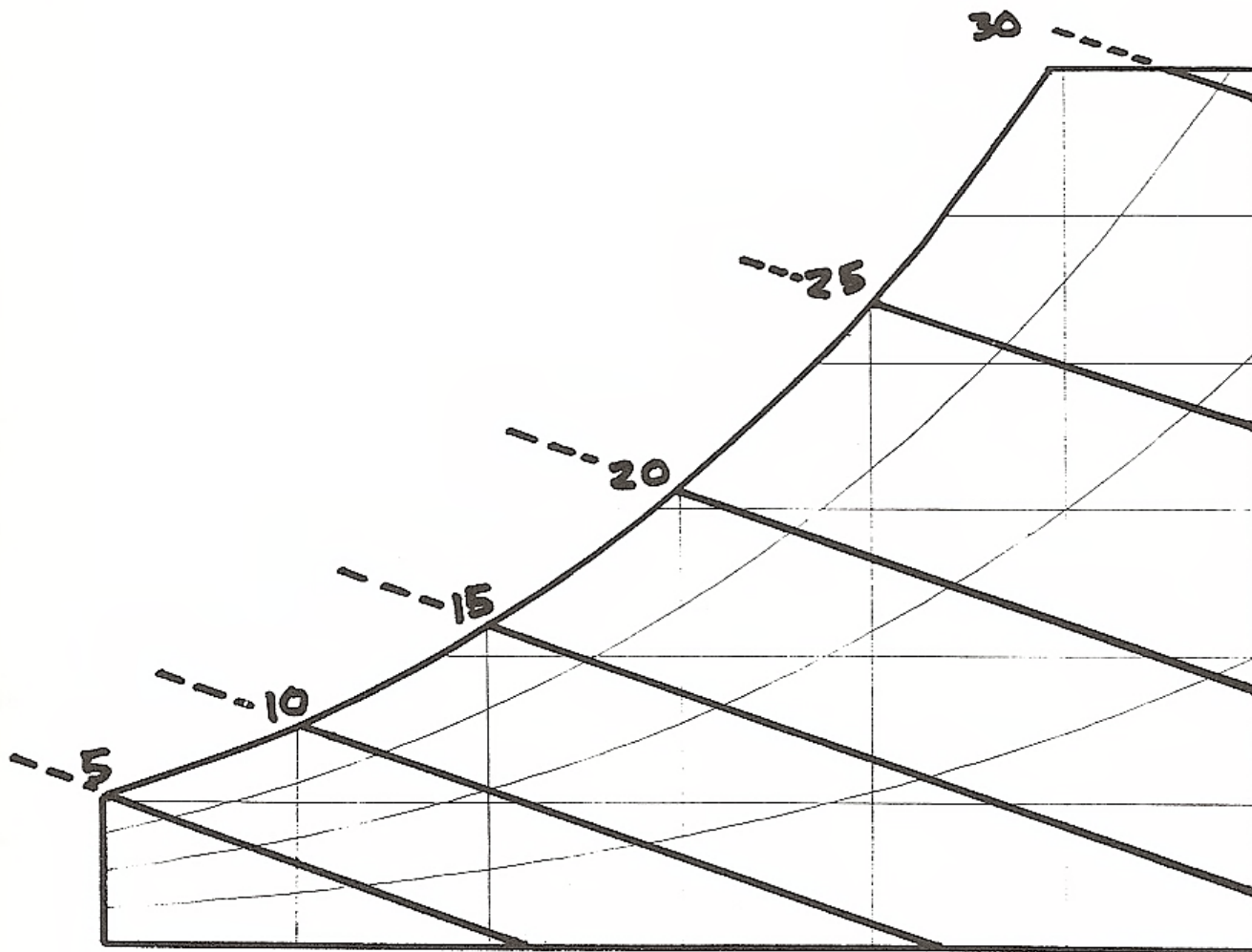


Figure 11-3 Wet-bulb temperature isolines (t_{WB} in $^{\circ}C$).

Source: Gatley, Understanding Psychrometrics, Chapter 11

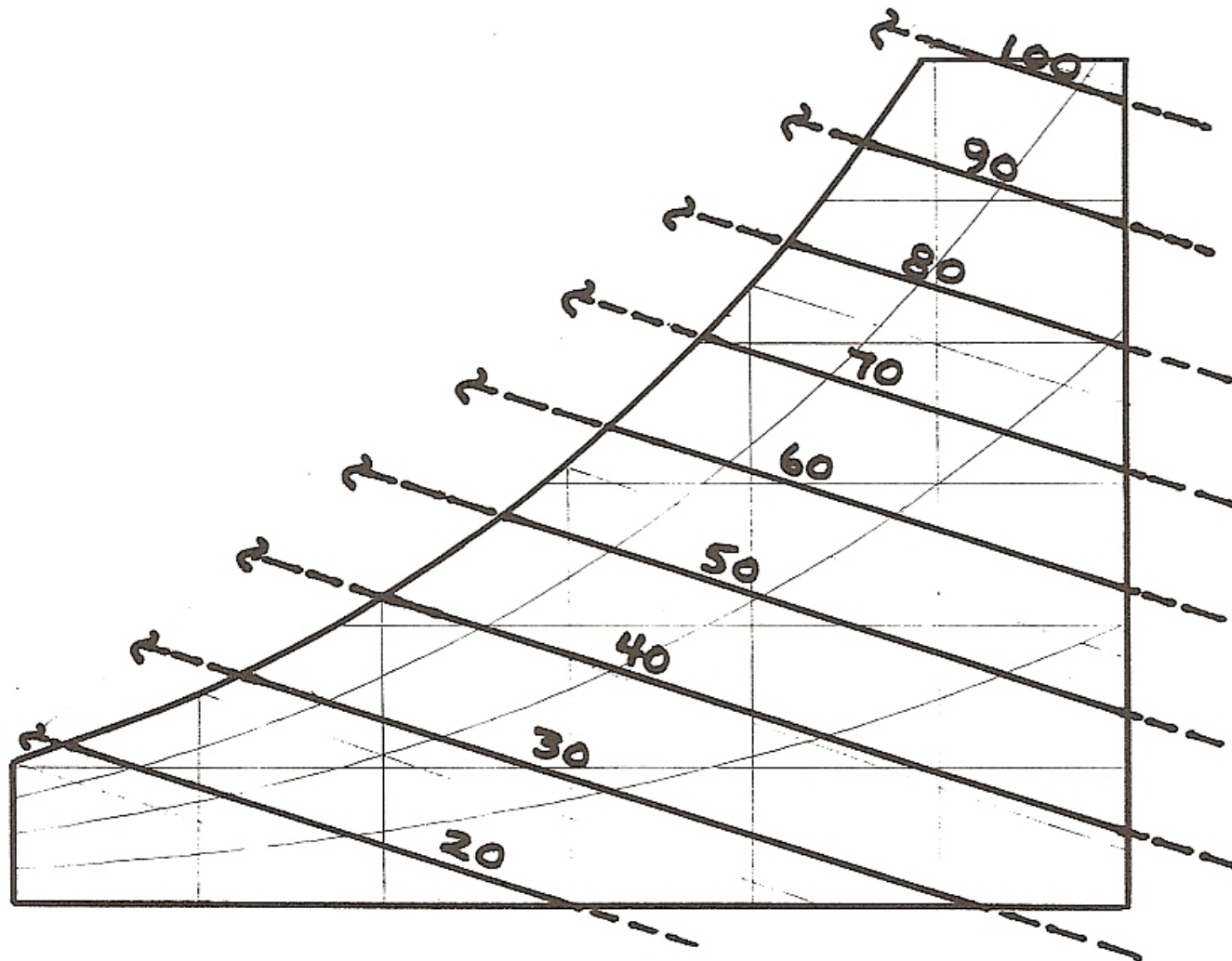


Figure 15-1 *Specific enthalpy isolines (h in units of $\text{kJ}/\text{kg}_{\text{DA}}$).*

Source: Gatley, Understanding Psychrometrics, Chapter 15

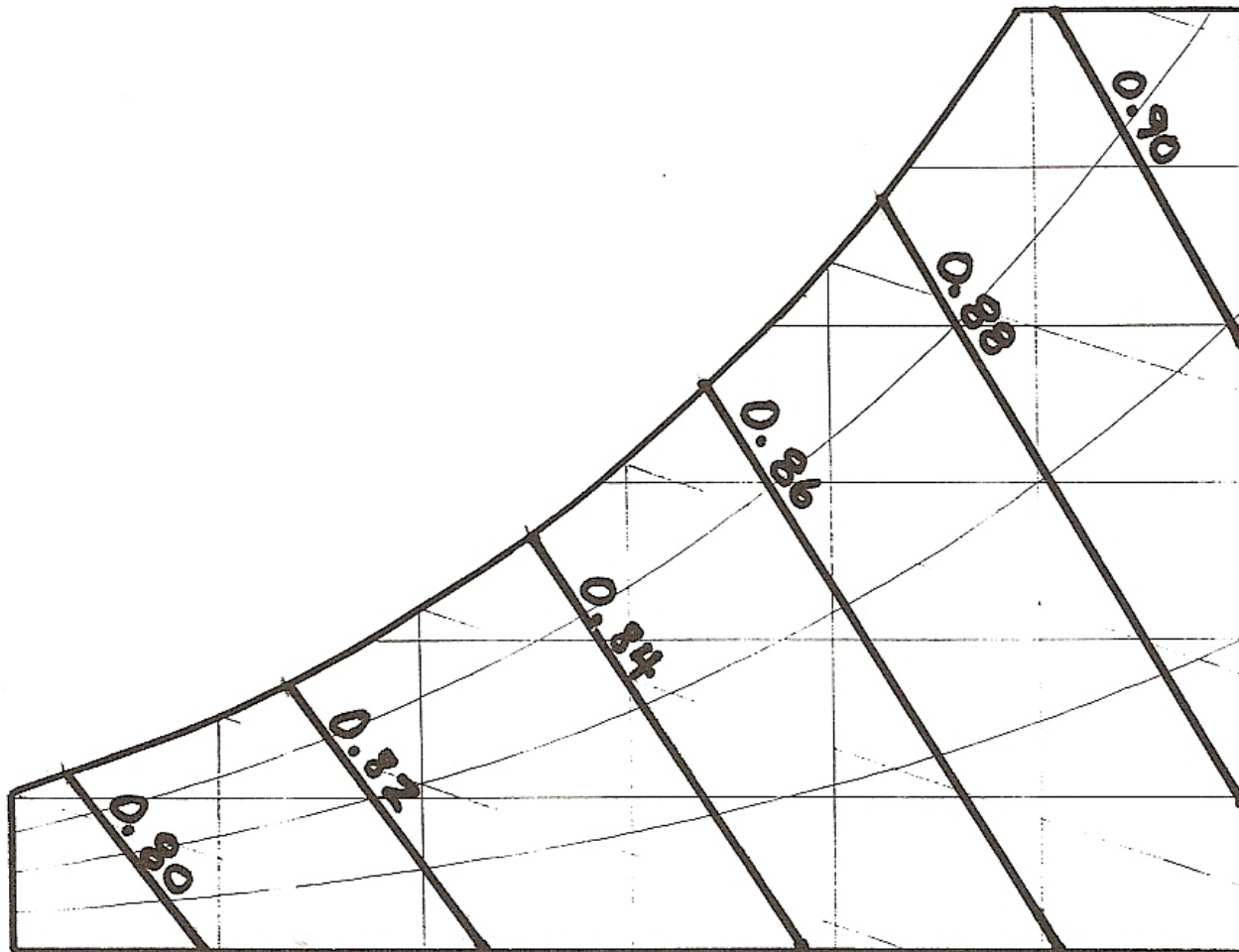
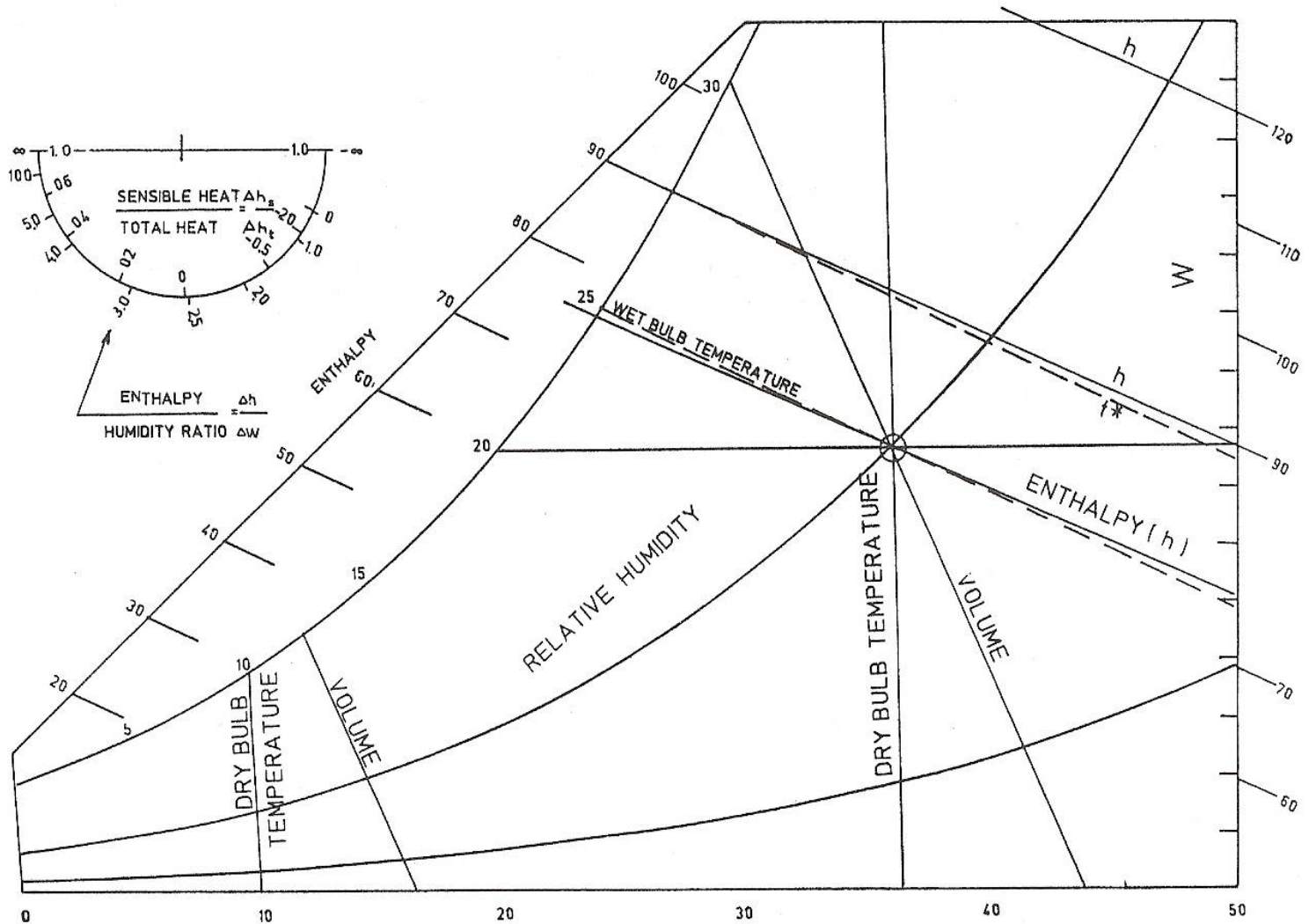


Figure 16-1 *Specific volume isolines (v in units of m^3/kg_{DA}).*

Source, Gatley, Understanding Psychrometrics, Chapter 16

Figure A4-10 Skeleton view of No. 1 SI—Normal Temperature—Sea Level.



Source: Gatley, Understanding Psychrometrics, Appendix IV

Pyschrometrics

To find a “state point” you need to know three properties

- Pressure => usually barometric pressure
- Dry-bulb temperature
- A moisture property
 - wet-bulb temperature
 - relative humidity
 - dewpoint temperature

$$p_{\text{BAR}} = 101.325 \text{ kPa}$$

One Statepoint,
Saturation Curve, and
Five Property Isolines Shown -
Specific Enthalpy and Specific
Volume Isolines Not Shown

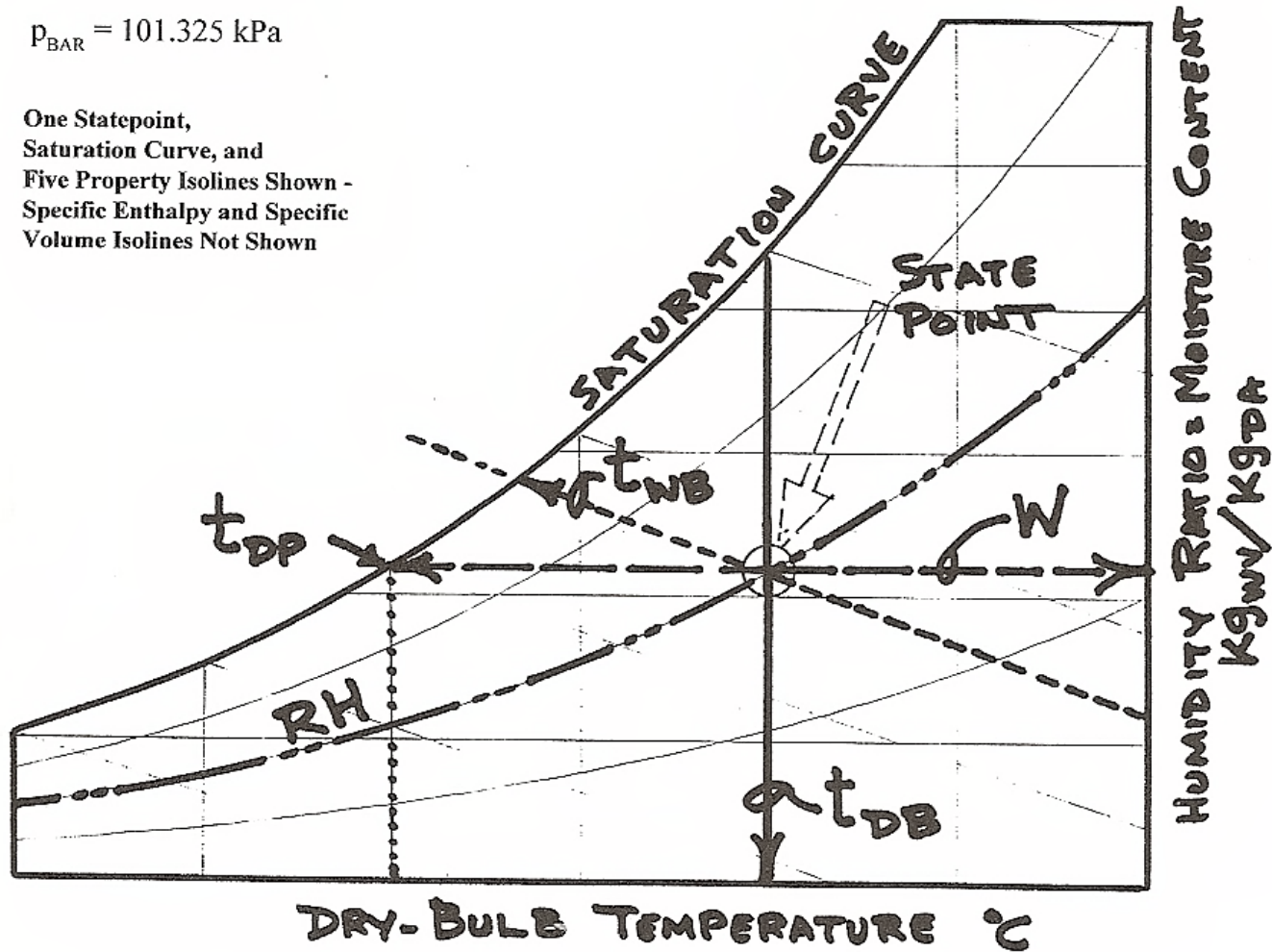


Figure 9-1 Basic psychrometric chart.

Source: Gatley, Understanding Psychrometrics, Chapter 9

Pyschrometrics

Four Basic Processes

- Sensible heating
- Sensible cooling
- Humidifying
- Dehumidifying only

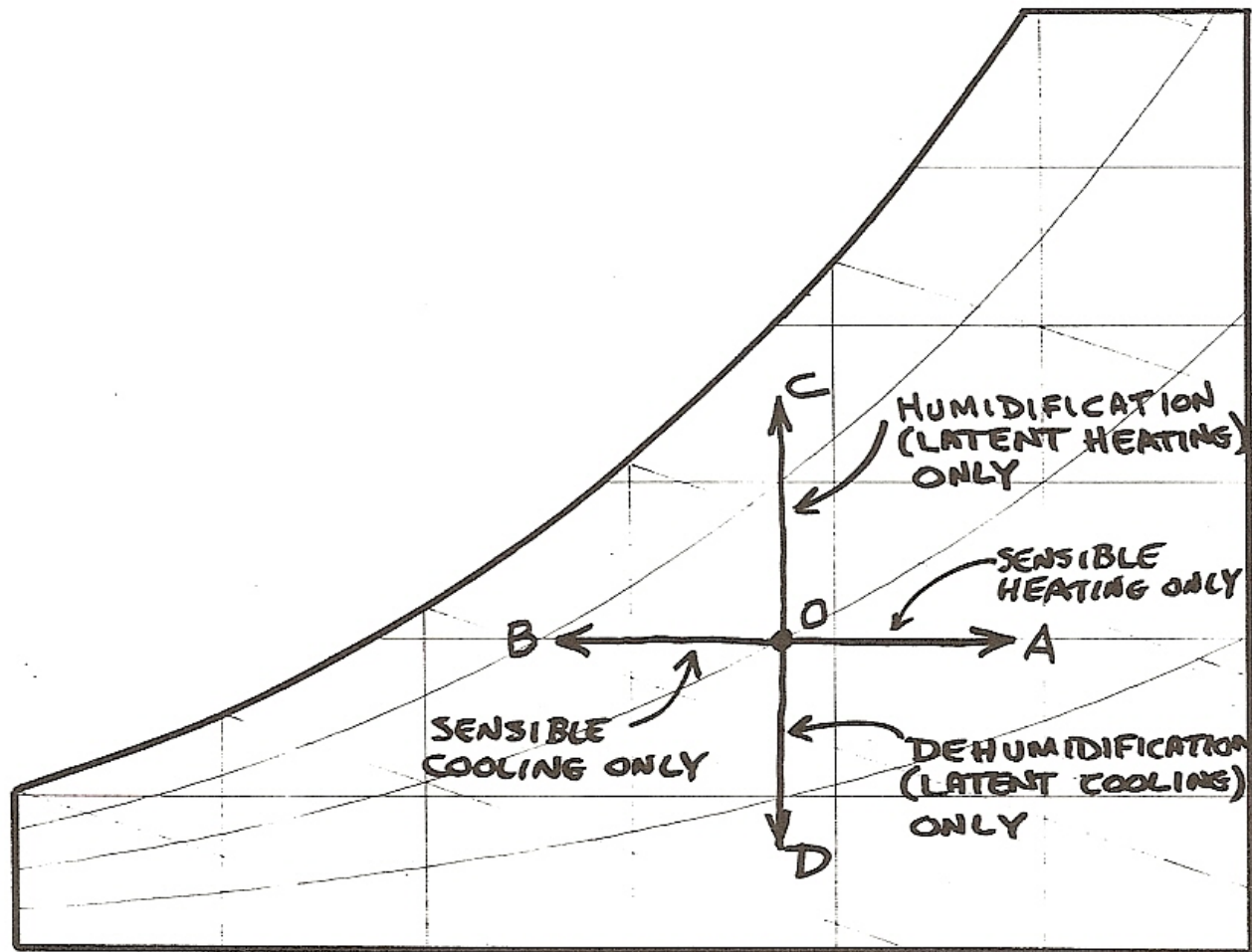


Figure 19-1 *Four basic processes.*

Source: Gatley, Understanding Psychrometrics, Chapter 19

Pyschrometrics

Other Common Processes

- Chemical dehumidifying
- Evaporative cooling
- Cooling and dehumidifying
- Heating and humidifying

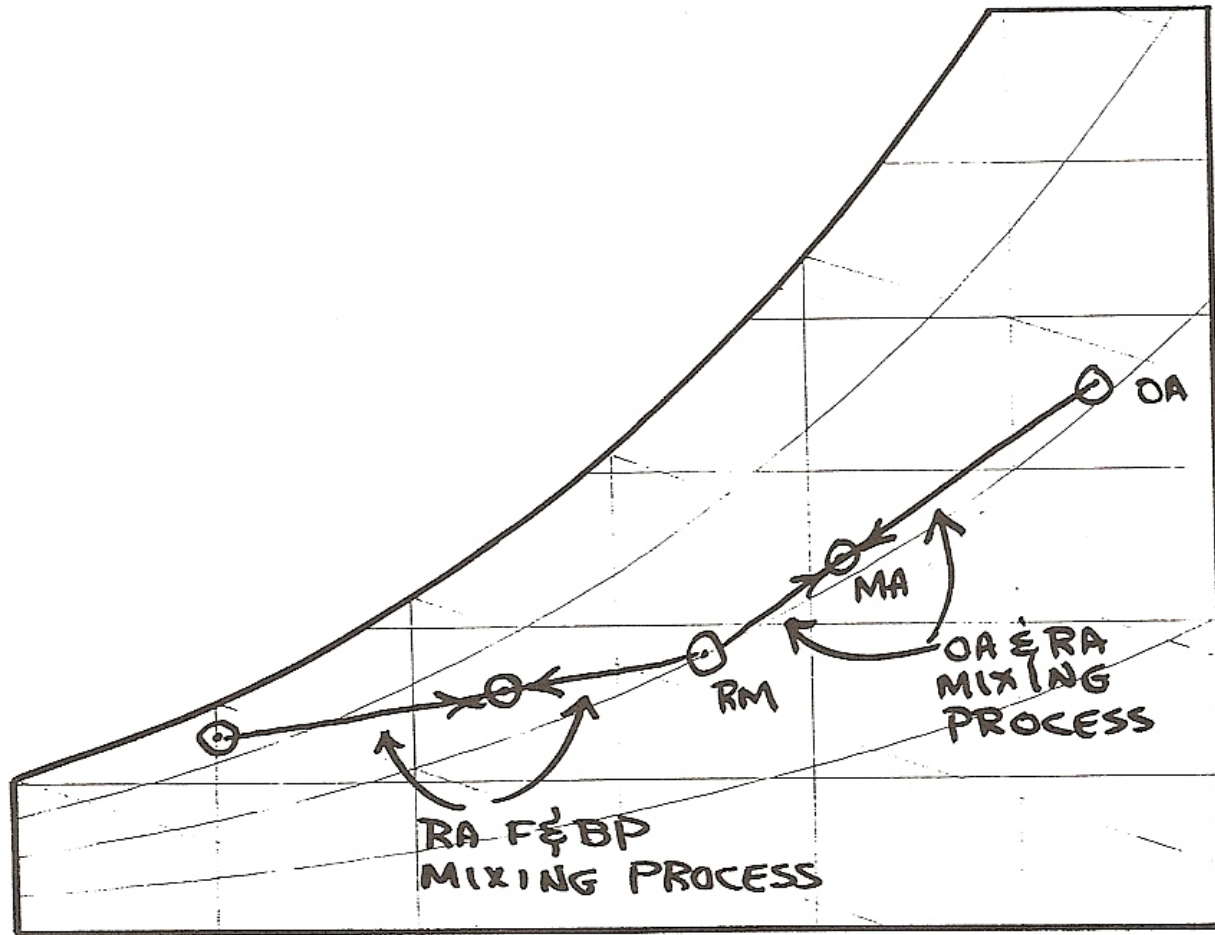


Figure 20-1 *Mixing processes.*

Source: Gatley, *Understanding Psychrometrics*, Chapter 20

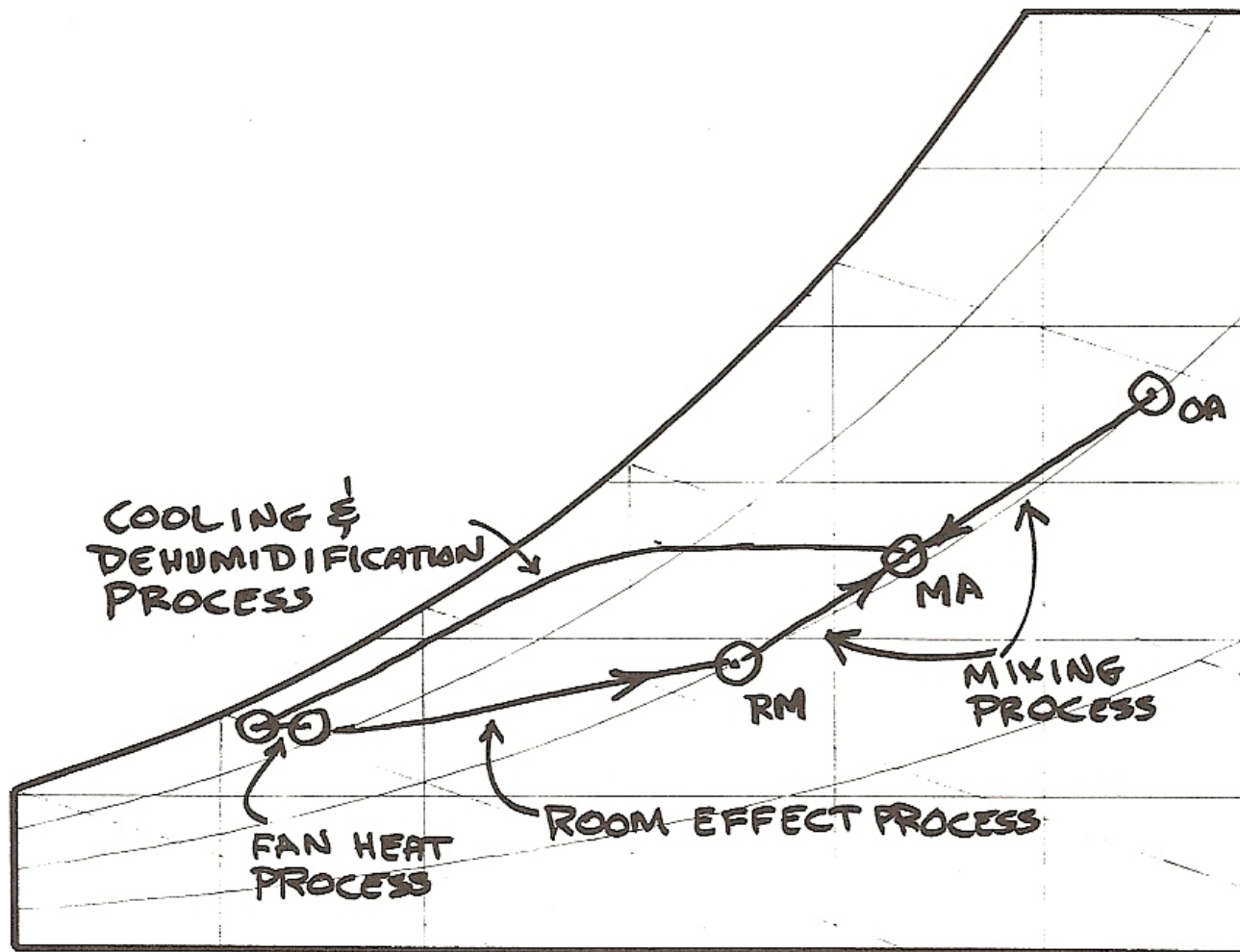
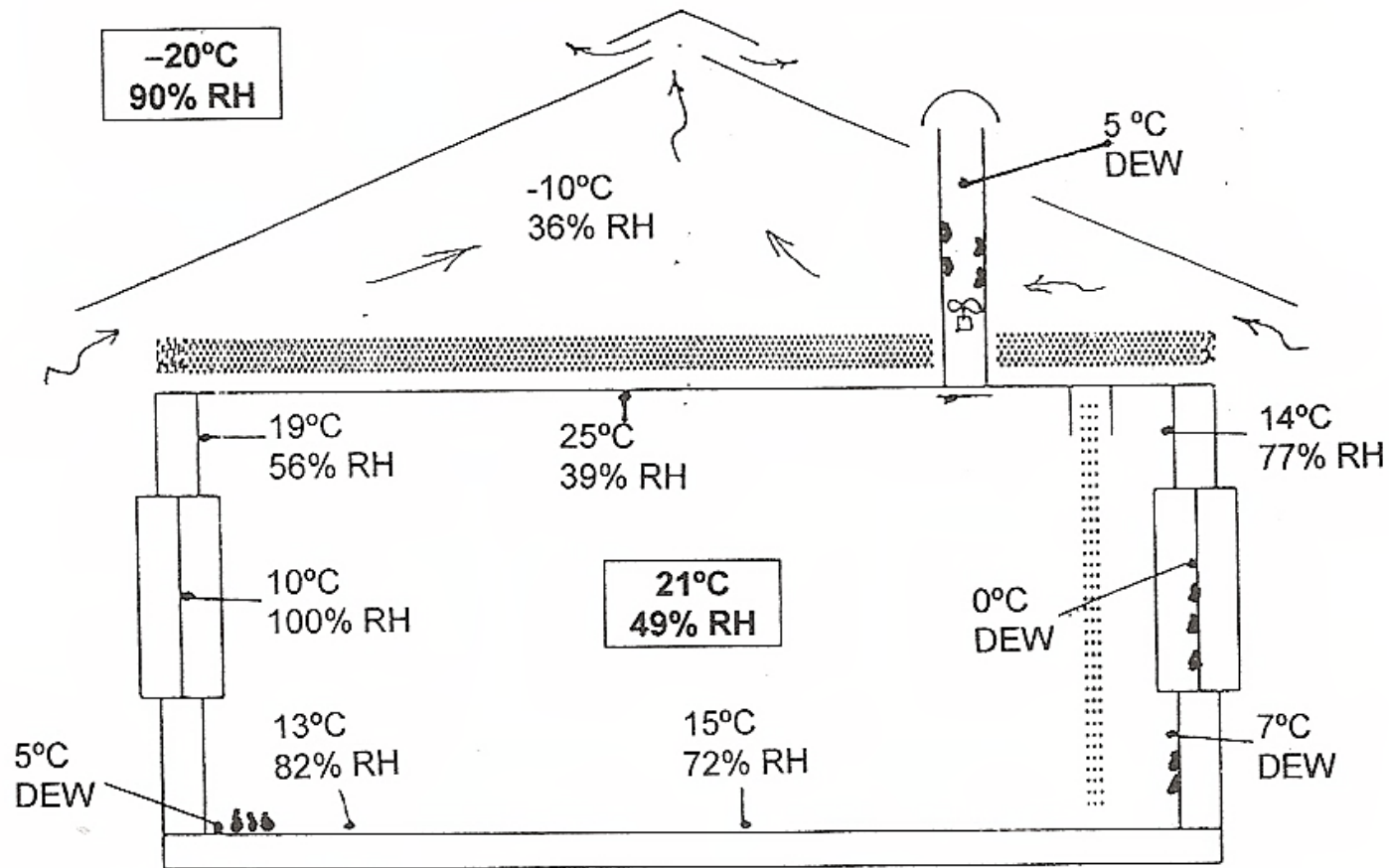


Figure 19-3 A four-process air-conditioning cycle.

Source: Gatley, Understanding Psychrometrics, Chapter 19



ROOM_{CTR}: 21°CDB, 10°C DP, 49% RH, p_{wv} 1228 Pa (0.3626 "Hg), W 0.00766 kg_{wv} / kg_{DA}
 OUTSIDE: -20°CDB, -21°C DP, 90% RH, p_{wv} 93 Pa (0.0275 "Hg), W 0.00058 kg_{wv} / kg_{DA}

Figure 13-5 Inside relative humidities in cold weather.

Source: Gatley, Understanding Psychrometrics, Chapter 13

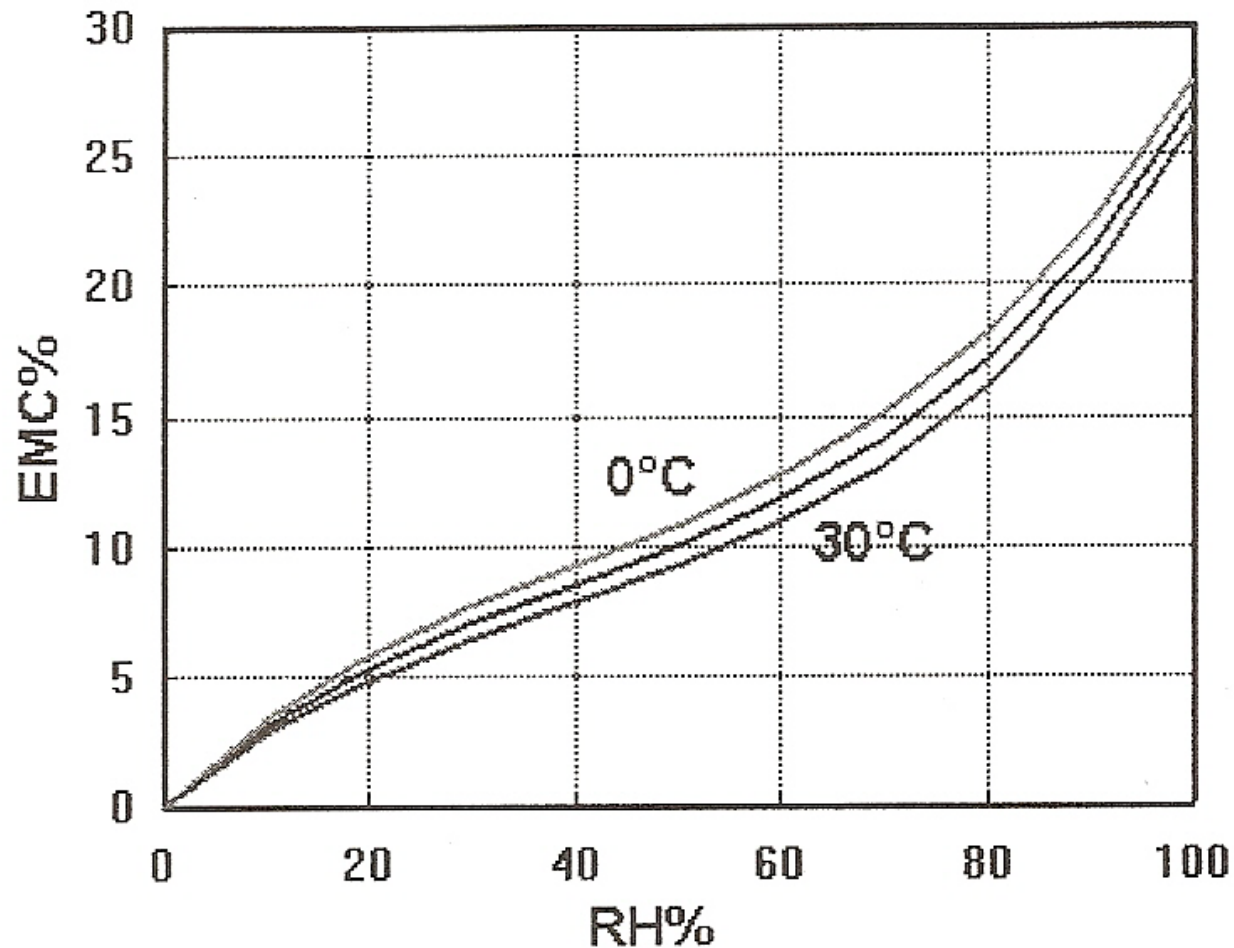


Figure 13-1 *Equilibrium moisture content of wood vs. relative humidity (by permission of Tim Padfield, Ph.D.; www.natmus.min.dk/cons/tp).*

Source: Gatley, Understanding Psychrometrics, Chapter 13

In Summary

Questions and Discussion

Next Class

- Moisture Transport & Control in Insulated Assemblies
 - General moisture concerns
 - Moisture in building materials
 - Moisture migration
 - Vapor diffusion vs. convective mass transport
- Readings
 - HF: Chapter 25 => 25.10 to 25.17
 - HF: Chapter 26 => 26.13 to 26.20
 - HF: Chapter 27 => 27.7 to 27.12
 - HPE: Chapter 3.5 & 3.6
 - BG: Pages 115 to 129 (Appendix II & III for editions prior to 2004)